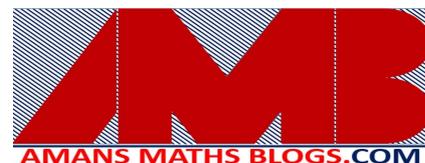


## JSTSE : PREVIOUS YEARS

### (CHEMISTRY : 2012)

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41. Which chemical is mixed in LPG to detect its leakage:
- (1) H<sub>2</sub>S (2) MIC  
(3) Mustard gas (4) Ethanethiol
42. Chemically the water gas is:
- (1) H<sub>2</sub>O (gaseous) (2) CO + H<sub>2</sub>  
(3) CO<sub>2</sub> + H<sub>2</sub> (4) CO + N<sub>2</sub>
43. The composition of Germansilver alloy is:
- (1) 70% Cu + 30% Zn (2) 60% Cu + 30% Zn + 10% Ag  
(3) 60% Ag + 40% Cu (4) 20% Cu + 30% Zn + 40% Ni
44. The Latin name of metal Tungsten is:
- (1) Natrium (2) Ferrym  
(3) Wolfram (4) Kalium
45. Which is not related to colloidal solution :
- (1) Tyndall Effect (2) Electrophoresis  
(3) Very unstable (4) Brownian movement
46. One amu equals to:
- (1)  $1.66 \times 10^{-24}$  kg (2)  $1.66 \times 10^{-24}$  g  
(3)  $1.66 \times 10^{-24}$  mg (4)  $1.66 \times 10^{-24}$  pounds
47. Umber of neutrons in 8 g He
- (1)  $6.022 \times 10^{23}$  (2)  $2 \times 6.022 \times 10^{23}$   
(3)  $3 \times 6.022 \times 10^{23}$  (4)  $4 \times 6.022 \times 10^{23}$
48. Who discovered the nucleus
- (1) J. J. Thomson (2) Goldstein  
(3) Rutherford (4) Chadwick



49. Cinnabar is an ore of:
- (1) Copper (2) Mercury  
(3) Silver (4) Zinc
50. Which cell is used in wrist watches:
- (1) Mercury cell (2) Daniel Cell  
(3) Leclanchecell (4) Fuel cell
51. Which rule is applicable on the dissolution of CO<sub>2</sub> gas in cold drinks:
- (1) Hess'Law (2) Henry's Law  
(3) Kohlrausch Law (4) Dalton's Law
52. Which one is not acidic in nature :
- (1) BF<sub>3</sub> (2) NH<sub>3</sub>  
(3) B(OH)<sub>3</sub> (4) AlCl<sub>3</sub>
53. Chemically the P.O.P. is:
- (1) CaSO<sub>4</sub>.2H<sub>2</sub>O (2) (CaSO<sub>4</sub>)<sub>2</sub>.H<sub>2</sub>O  
(3) Ca(OH)<sub>2</sub> (4) (CaSO<sub>4</sub>)<sub>2</sub>.H<sub>2</sub>O
54. Sodalime is:
- (1) NaOH + KOH (2) CaO + Ca(OH)<sub>2</sub>  
(3) NaHCO<sub>3</sub> + Na<sub>2</sub>CO<sub>3</sub> (4) NaOH + CaO
55. Which is not an alum:
- (1) K<sub>2</sub>SO<sub>4</sub>.Al<sub>2</sub>.(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O (2) Na<sub>2</sub>SO<sub>4</sub>.Al<sub>2</sub>.(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O  
(3) K<sub>2</sub>SO<sub>4</sub>.Cr<sub>2</sub>.(SO<sub>4</sub>)<sub>3</sub>.24H<sub>2</sub>O (4) FeSO<sub>4</sub>.(NH<sub>4</sub>)<sub>2</sub>.SO<sub>4</sub>.24 H<sub>2</sub>O
56. Metal present in chloroplast is:
- (1) Iron (2) Copper  
(3) Cobalt (4) Magnesium
57. Which water soluble vitamin can be stored in our body:
- (1) Vit-B<sub>1</sub> (2) Vit-B<sub>2</sub>  
(3) Vit-B<sub>6</sub> (4) Vit-B<sub>12</sub>
58. The main component of LPG is:
- (1) Methane (2) Ethane  
(3) Propane (4) Butane
59. The most abundant element of the universe is:
- (1) Oxygen (2) Silicon  
(3) Hydrogen (4) Sodium
60. BHA and BHT are used as:

- (1) Insecticide (2) Food Preservative  
(3) Drugs (4) Fuels
61. TEL is used as:  
(1) Telephone device (2) Medicine  
(3) Antiknock agent (4) Food Preservative
62. BOD and COD are:  
(1) Research Organisation  
(2) Organic compounds  
(3) Parameters to measure air pollution  
(4) Parameters to measure water pollution
63. The energy of an electron in an atom is:  
(1) Always negative (2) Always positive  
(3) Always zero (4) All the above
64. Tear gas is :  
(1)  $N_2O$  (2)  $O_2N$ .  $CCl_3$   
(3)  $SO_2$  (4)  $H_2S$
65. Which one is not a green house gas:  
(1)  $N_2O$  (2)  $CH_4$   
(3)  $H_2O$  (Vapour) (4)  $CO_2$
66. 2,4-Dis:  
(1) Fungicide (2) Insecticide  
(3) Bactericide (4) Weedicide
67. The pH of acid rain water is:  
(1)  $< 5.6$  (2)  $< 7.0$   
(3)  $< 8.0$  (4)  $> 7.0$
68. The name of a polymeric plastic on which oil and water do not stick and is used to make non-sticky cook-wares is:  
(1) Polythene (2) Bakelite  
(3) Dacron (4) Teflon
69. The chemical composition of rust of copper (green coating) is:  
(1)  $CuSO_4 \cdot 5H_2O$  (2)  $Cu(OH)_2 + CuCO_3$   
(3)  $CuSO_4 + Fe_2O_3$  (4)  $CuSO_4 + CuCO_3$
70. Choose the incorrect pair:  
(1)  $MgO$  - Basic Oxide (2)  $SO_2$  -Acidic Oxide  
(3)  $CO$  - Neutral Oxide (4)  $P_4O_{10}$  - Basic Oxide

71. Bitumen is:
- (1) a petroleum product used in place of coal tar
  - (2) an insecticide used to kill bests
  - (3) a fossil fuel used to burn in heavy vehicle
  - (4) a polymer used to make cloths
72. What is gasohol?
- (1) Green house gas
  - (2) a hole in space
  - (3) mixture of petrol and ethyl alcohol
  - (4) mixture of petrol & kerosene oil
73. Which one has highest calorific value?
- (1) CNG
  - (2) LPG
  - (3) Hydrogen Gas
  - (4) Biogas
74. Choose incorrect pair:
- (1) Hair-protein
  - (2) Bones-calcium phosphate
  - (3) Cotton - Carbohydrate
  - (4) Silk-hat
75. Which hormone is used for 'induced lactation'?
- (1) Adrenaline
  - (2) Oxytocin
  - (3) Vasopressin
  - (4) Auxin
76. The heat energy in sun/stars is produced by:
- (1) Nuclear fission reaction
  - (2) Combustion reaction
  - (3) Nuclear fusion reaction
  - (4) Thermochemical reaction
77. What is wrong about fatty acids:
- (1) a component of fat
  - (2) found in fatty men only
  - (3) always contain even number of carb-atom
  - (4) used to make soaps
78. Which of the following will not sublime:
- (1) camphor
  - (2) Ammonium chloride
  - (3) Calcium Chloride
  - (4) Iodine
79. The temperature at which vapour pressure of a liquid becomes equal to the atmospheric pressure is:
- (1) Boiling point
  - (2) Melting point
  - (3) Freezing point
  - (4) Sublimation point
80. In which sodium metal cannot be stored:
- (1) Kerosenoil
  - (2) Paraffin wax
  - (3) Ether
  - (4) Alcohol